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Listing of claims

- 1. (Currently amended) A process to prepare nano-size Zeolite A comprising the steps of:
 - a. preparing a precursor mixture comprising sodium, silica, and alumina, wherein the proportions of sodium, silica, and alumina are those required to produce Zeolite A, to form a solid precursor;
 - b. isolating the solid precursor from step a);
 - c. combining the solid precursor from step b) with seed crystals of Zeolite A having particle sizes of less than 500nm, with agitation in an aqueous alkaline solution containing hydroxide ions at a temperature of about 18°C to about 85°C to form a mixture containing nano-sized Zeolite A crystals; and
 - d. optionally separating the nano-sized Zeolite A crystals from the mixture of step c).
 - 2. (Original) The process as of Claim 1 wherein the precursor mixture is a non-clear solution.
 - 3. (Original) The process as of Claim 2 wherein the precursor mixture is an amorphous aluminosilicate gel.
 - 4. (Original) The process as of Claim 3 wherein the precursor mixture is an amorphous aluminosilicate gel prepared from NaAlO2, NaOH, and tetraethoxysilane.
 - 5. (Original) The process as of Claim 3 wherein the amorphous aluminosilicate gel is prepared at a temperature of about 70°C to about 100°C.
 - 6. (Original) The process as of Claim 1 wherein the aqueous alkaline solution is an aqueous solution of (CH₃)₄NOH or NaOH.
 - 7. (Original) The process as of Claim 6 wherein the aqueous solution of (CH₃)₄NOH or NaOH is at a concentration of about 0.5 to about 3.0 molar.
 - 8. (Cancelled)
 - 9. (Original) The process as of Claim § 1 wherein the seed crystals of Zeolite A are particles of less than 250nm.
 - 10. (Original) The process as of Claim 1 wherein the mixture formed in step c) is agitated for about 1 day to about 20 days before proceeding to step d).
 - 11. (Original) The process as of Claim 10 wherein the solution formed in step c) is agitated for more than 10 days.
 - 12. (Original) A process to prepare nano-size Zeolite A comprising the steps of:

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- a. preparing a precursor mixture comprising sodium, silica, and alumina, wherein the proportions of sodium, silica, and alumina are those required to produce Zeolite A, to form a solid precursor;
- b. isolating the solid precursor from step a);
- c. combining the solid precursor from step b) with seed crystals of Zeolite A with agitation in an aqueous alkaline solution containing hydroxide ions at a temperature of about 18°C to about 85°C to form a mixture containing nano-sized Zeolite A crystals;
- d. optionally separating the nano-sized Zeolite A crystals from the mixture of step c);
- e. adding the nano-sized Zeolite A crystals with agitation from step d) to an aqueous solution of a Na+ salt to form a mixture containing nanosized Zeolite A crystals; and
- f. optionally isolating the nano-sized Zeolite A crystals.
- 13. (Original) The process as in Claim 12 wherein the mixture formed in step e) is agitated for about 1 day to about 20 days before proceeding to step f).
- 14. (Original) The process as of Claim 1 wherein the aqueous alkaline solution additionally contains one or more of a C₁-C₆ acetone or alcohol.
- 15. (Original) The process as of Claim 14 wherein aqueous alkaline solution additionally contains ethanol.
- 16. (Currenty amended) A <u>product comprising</u> of nano-sized Zeolite A crystal <u>particles</u> prepared by the process of Claim 1 <u>wherein</u> 95% of particles have a particle <u>size distribution of 50 to 150 nm.</u>
- 17. (Original) Non-phosphate detergent builders, thin films, catalysts, and micro-patterns containing the nano-sized Zeolite A of Claim 16.